

Date Planned : / /	Daily Tutorial Sheet-1	Expected Duration : 90 Min		
Actual Date of Attempt : / /	JEE Main (Archive)	Exact Duration :		

1.	When r (A) (B) (C) (D)	rain is accompanied by a thunderstorm, the collected rain water will have a pH value: which depends upon the amount of dust in air slightly lower than that when rain water without thunderstorm slightly higher than that when the thunderstorm is not there uninfluenced by occurrence of thunderstorm									
2.	Smog is essentially caused by the presence of :								(2004)		
	(A)	$O_3$ and $N_2$	7	AT.	(B)	$O_2$ and $N_2$	=				
0	(C)	oxides of Sulph		_	<b>(D)</b>	O <sub>2</sub> and O <sub>3</sub>	3		(2008)		
3.	(A) (B) (C) (D)	fy the wrong statement among the following:  chlorofluorocarbons are responsible for ozone layer depletion  greenhouse effect is responsible for global warming  ozone layer does not permit infrared radiation from the sum to reach the earth.  acid rain is mostly because of oxides of nitrogen and sulphur									
4.		one of the follow	•			1			(2011)		
	(A) (B) (C) (D)	oxides of sulphur, nitrogen and carbon are the most widespread air pollutant.  pH of drinking water should be between 5.5 and 9.5  concentration of BOD below 6 ppm is good for the growth of fish  clean water would have a BOD value of less than 5 ppm									
5.							og is not co	rrect ?	(2012)		
	(A) (B) (C) (D)	photochemical smog is formed through photochemical reaction involving solar energy photochemical smog does not cause irritation in eyes and throat carbon monoxide does not play any role in photochemical smog reaction photochemical smog is an oxidizing agent in character									
6.	The gas leaked from a storage tank of the Union Carbide plant in Bhopal gas tragedy was :								(2013)		
	(A) (C)	ammonia methylisocyanate			(B) (D)	phosgene methylamine					
7.	nitrile (	Photochemical smog consists of excessive amount of X, in addition to aldehydes, ketones, peroxy nitrile (PAN), and so forth. X is :									
	(A)	$CH_4$	(B)	СО	(C)	$CO_2$	(D)	$O_3$			
8.	Addition (A) (C)	n of phosphate f Enhanced grow Deposition of ca	th of alg	ae	(B) (D)	Increase in		dissolved oxyg ation	(2015) en in water		
9.	found t	concentration of fluoride, lead, nitrate and iron in a water sample from an underground lake was d to be 1000 ppb, 40 ppb, 100 ppm and 0.2 ppm, respectively. This water is unsuitable for drinking to high concentration of:  (2016)									
	(A)	Lead	<b>(B)</b>	Nitrate	(C)	Iron	<b>(D)</b>	Fluoride			
10.	BOD stands for :							(2016)			
	(A)					Bacterial Oxidation Demand					
	(C) Biochemical Oxygen Demand					Biochemical Oxidation Demand					